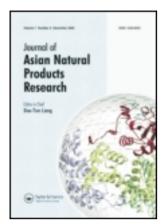
This article was downloaded by: [Malmo Hogskola]

On: 20 December 2011, At: 23:15

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered

office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/ganp20

Two new phenylpropanoids and one propanoate from Morinda citrifolia

Gang Wang ^a , Qi-Wei He ^{a b} , Tao Feng ^b & Ji-Kai Liu ^b

^a Anhui Key Laboratory of Modernized Chinese Materia Medica, Anhui University of Traditional Chinese Medicine, Hefei, 230031, China

^b State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming, 650204, China

Available online: 15 Mar 2011

To cite this article: Gang Wang, Qi-Wei He, Tao Feng & Ji-Kai Liu (2011): Two new phenylpropanoids and one propanoate from Morinda citrifolia, Journal of Asian Natural Products Research, 13:03, 238-241

To link to this article: http://dx.doi.org/10.1080/10286020.2010.551344

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.tandfonline.com/page/terms-and-conditions

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



Two new phenylpropanoids and one propanoate from *Morinda* citrifolia

Gang Wang^a, Qi-Wei He^{ab}, Tao Feng^b and Ji-Kai Liu^b*

^aAnhui Key Laboratory of Modernized Chinese Materia Medica, Anhui University of Traditional Chinese Medicine, Hefei 230031, China; ^bState Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming 650204, China

(Received 10 November 2010; final version received 26 December 2010)

Two new phenylpropanoids, methyl 3-(2,4-dihydroxy-5-methoxyphenyl)propionate (1) and butyl 3-(2,4-dihydroxy-5-methoxyphenyl)propionate (2), and one unusual propanoate, 5-hydroxyhexyl 2-hydroxypropanoate (3), were isolated from the fruits of *Morinda citrifolia*. Their structures were established using MS and NMR methods.

Keywords: Morinda citrifolia; Rubiaceae; phenylpropanoids; propanoate

1. Introduction

Morinda citrifolia L. (Rubiaceae), known as 'noni', is a small tree that grows widely across Polynesia [1]. There is a long history of the use of M. citrifolia as an important medicinal plant for the treatment of asthma, bone fractures, cancer, cholecystitis, dysentery, lumbago, menstrual cramps, urinary difficulties, and many other ailments [2]. Recent phytochemical studies on this plant have reported anthraquinones and anthraquinone glycosides, fatty acid glycosides, iridoids and iridoid glycosides, lignans, and triterpenoids. Of them, the novel iridoids including citrifolinin A [3], citrifolinoside [4], yopaaoside A [5], yopaaoside B [5], and morindacin [6] were isolated. They showed activities in a wide range of biological assays such as the inhibition of angiogenesis, 5-lipoxygenase and 15-lipoxygenase, cyclooxygenase-1, phorbol ester-induced inflammation and tyrosine kinase [7–11]. Here, we report the isolation of three new compounds, methyl 3-(2,4-dihydroxy-5-methoxyphenyl)propionate (1), butyl 3-(2,4-dihydroxy-5-methoxyphenyl)propionate (2), and 5-hydroxyhexyl 2-hydroxypropanoate (3) (Figure 1), from the EtOAc and *n*-BuOH extract of the fruits of *M. citrifolia*.

2. Results and discussion

Compound 1 was isolated as a yellow gum. The molecular formula $C_{11}H_{14}O_5$ was established by the positive HR-ESI-MS at m/z 249.0745 [M + Na]⁺. In the ¹H NMR spectrum, two singlets at $\delta_{\rm H}$ 3.81 and 3.68 (each 3H) were assigned to two methoxyl groups, two singlets at $\delta_{\rm H}$ 6.55 and 6.52 (each 1H) were assigned to two aromatic protons, and four protons at $\delta_{\rm H}$ 2.82 and 2.68 (each 2H, t, $J=6.4\,{\rm Hz}$) were ascribed to protons of two conjoint methylenes. The ¹³C NMR spectrum displayed 11 carbon resonances ascribable to 2 methoxyl groups

^{*}Corresponding author. Email: jkliu@mail.kib.ac.cn

Figure 1. Chemical structures of compounds 1-3.

 $(\delta_{\rm C} 56.5, 52.3), 2$ methylenes $(\delta_{\rm C} 23.5,$ 29.4), 2 methines ($\delta_{\rm C}$ 104.0, 109.6), and 5 quaternary carbons ($\delta_{\rm C}$ 113.0, 138.6, 144.5, 147.4, 169.3). These NMR spectral data suggested that compound 1 was an α,β -saturated phenylpropanoid. One of the methoxyl groups was placed at C-5 according to the HMBC correlation between 5-OCH₃ and C-5, and the ROESY correlation of H-6 with 5-OCH₃ (Figure 2). In addition, the HMBC correlation of the methoxyl at $\delta_{\rm H}$ 3.68 (3H, $-OCH_3$) with C-3' at δ_C 169.3 suggested a methyl ester group (Figure 2). Detailed analysis of HMBC and ROESY spectra (Figure 2) established the structure of compound 1 as methyl 3-(2,4-dihydroxy-5-methoxyphenyl)propionate.

Compound **2** was isolated as a yellow gum. The HR-ESI-MS established the molecular formula $C_{14}H_{20}O_5$ (m/z 291.1201 [M + Na]⁺, calcd 291.1208). All the NMR spectral data suggested that compound **2** had a similar structure to that of **1** except for a n-butyl ester in **2** instead of the methyl ester in **1** as suggested by the HMBC correlations of the oxymethylene at

 $\delta_{\rm H}$ 4.08 (2H, t, $J=6.5\,{\rm Hz}$) with C-3' at $\delta_{\rm C}$ 175.4 and the MS data. Detailed analysis of 2D NMR (HSQC, HMBC, $^1{\rm H}-^1{\rm H}$ COSY) spectra finally established the structure of compound 2 as butyl 3-(2,4-dihydroxy-5-methoxyphenyl)propionate. However, compound 2 might be an artificial product due to the usage of n-BuOH in the extraction.

Compound 3 was obtained as a colorless oil. Its molecular formula C₉H₁₈O₄ was determined by the negative HR-ESI-MS at m/z 189.1127 [M-H]⁻. The IR spectrum revealed the presence of hydroxyl (3389 cm⁻¹) and carbonyl (1736 cm⁻¹) groups. The ¹³C NMR spectrum displayed nine carbon resonances including one carbonyl group ($\delta_{\rm C}$ 175.8), two methines ($\delta_{\rm C}$ 66.7, 67.8), four methylenes ($\delta_{\rm C}$ 22.0, 28.5, 38.6, 65.5), and two methyl ($\delta_{\rm C}$ 20.4, 23.6) carbons. Correspondingly, the ¹H NMR spectrum revealed two methyl groups at δ_H 1.19, $(d, J = 6.2 \,Hz)$ and $1.41 \,(d, J = 6.2 \,Hz)$, two oxymethines at $\delta_{\rm H}$ 4.27 (m) and 3.79 (m), one oxymethylene at $\delta_{\rm H}$ 4.19 (m), and three methylenes at $\delta_{\rm H}$ 1.44–1.68.

Figure 2. Key 2D NMR correlations of 1 and 3.

The $^{1}H^{-1}H$ COSY spectrum revealed two partial fractions (**a** CH₃CH and **b** CH₃CHCH₂CH₂CH₂CH₂CH₂) (Figure 2). In the HMBC spectrum, the key correlation of an oxymethylene at $\delta_{\rm H}$ 4.17 (2H, m, H-1') with C-1 at $\delta_{\rm C}$ 175.8 suggested the connection of partial fractions **a** and **b** via through an ester group (Figure 2). Thus, the planar structure of **3** was determined as 5-hydroxyhexyl 2-hydroxypropanoate. Unfortunately, the stereo configuration at C-2 and C-5' could not be identified due to the limited quantity available.

3. Experimental

3.1 General experimental procedures

Optical rotations were measured with a Horiba SEPA-300 polarimeter. UV spectra were obtained using a Shimadzu UV-2401A spectrophotometer. A Tenor 27 spectrophotometer was used for scanning IR spectroscopy using KBr pellets. 1D and 2D NMR spectra were run on Bruker DRX-500 and AM-400 spectrometers with TMS as the internal standard. Chemical shifts (δ) were expressed in ppm with reference to the solvent signals. HR-ESI-MS were performed on an API-Qstar-Pulsar-1 spectrometer. Column chromatography was performed on silica gel (200-300 mesh, Qingdao Haiyang Chemical Co. Ltd., Qingdao, China) and RP-18 gel (20-45 µm, Fuji Silysia Chemical Ltd., Aichi, Japan). Fractions were monitored by TLC (GF 254, Qingdao Haiyang Chemical Co. Ltd.), and spots were visualized by heating silica gel plates sprayed with 10% H₂SO₄ in EtOH.

3.2 Plant material

The dried powder of the fruit of *Morinda* citrifolia was obtained from Hsiehs Biotech., Vietnam in March 2008 and identified by Prof. Hua Peng, and a voucher specimen (YP08019) has been deposited in Hsiehs Biotech., Vietnam.

3.3 Extraction and isolation

Dried noni fruits (1100 g) were extracted with 95% EtOH (11) at room temperature for 1 week. The extract was concentrated to dryness under reduced pressure, and the residue was suspended in H₂O (1000 ml) and partitioned successively with EtOAc $(3 \times 500 \,\mathrm{ml})$ and *n*-BuOH $(3 \times 500 \,\mathrm{ml})$. The *n*-BuOH-soluble fraction (61 g) was subjected to column chromatography over silica gel (600 g, $140 \text{ cm} \times 7 \text{ cm}$; CHCl₃: $Me_2CO = 1:0 \rightarrow 1:1$) to give six subfractions 1-6. Subfraction 5 (4g) was separated by RP-18 (100 g, $60 \,\mathrm{cm} \times 4 \,\mathrm{cm}$; MeOH: $H_2O = 1:1$) to afford 1 (1.1 mg) and 2 (2.5 mg). The EtOAc fraction (21 g) was subjected to column chromatography on silica gel column chromatography (400 g, $60 \,\mathrm{cm} \times 4 \,\mathrm{cm}$; CHCl₃:Me₂- $CO = 1:0 \rightarrow 1:1$) to afford subfractions 6-9. Subfraction 9 (3 g) was separated by silica gel (100 g, $50 \,\mathrm{cm} \times 3 \,\mathrm{cm}$; petroleum ether: $Me_2CO = 2:1$) and further purified by Sephadex LH-20 (30 g, $150 \text{ cm} \times 2 \text{ cm}$; $CHCl_3:MeOH = 1:1$) to yield 3 (2 mg).

3.3.1 Methyl 3-(2,4-dihydroxy-5-methoxyphenyl)propionate (1)

Yellow gum; UV (MeOH) $λ_{max}$ (log ε): 204 (3.76), 290 (3.02) nm; IR (KBr) $ν_{max}$: 3439 (OH), 1737 (C=O) cm⁻¹; ¹H NMR (CDCl₃, 20°C, TMS) and ¹³C NMR (CDCl₃, 20°C, TMS) spectral data, see Table 1. HR MS ((+)ESI): m/z 249.0745 [M + Na]⁺ (calcd for C₁₁H₁₄O₅Na, 249.0738).

3.3.2 Butyl 3-(2,4-dihydroxy-5-methoxyphenyl)propionate (2)

Yellow gum; UV (MeOH) $λ_{max}$ (log ε): 204 (3.78), 290 (3.16) nm; IR (KBr) $ν_{max}$: 3441 (OH), 1737 (C=O) cm^{-1; 1}H NMR (CDCl₃, 20°C, TMS) and ¹³C NMR (CDCl₃, 20°C, TMS) spectral data, see Table 1. HR MS ((+)ESI): m/z 291.1201 (calcd for $C_{14}H_{20}O_5Na$, 291.1208).

Position	1		2	
	$\delta_{ m H}$	$\delta_{ m C}$	$\delta_{ m H}$	$\delta_{ m C}$
1		113.0 (s)		118.0 (s)
2		147.4 (s)		148.6 (s)
3	6.52 (1H, s)	104.0 (d)	6.52 (1H, s)	104.5 (d)
4		138.6 (s)		140.0 (s)
5		144.5 (s)		145.0 (s)
6	6.55 (1H, s)	109.6 (d)	6.55 (1H, s)	112.6 (d)
1'	2.82 (2H, t, 6.4)	23.5 (t)	2.81 (2H, t, 6.4)	24.2 (t)
2'	2.68 (2H, t, 6.4)	29.4 (t)	2.67 (2H, t, 6.4)	35.5 (t)
3'	, , , , , ,	169.3 (s)	, , , , , ,	175.4 (s)
5-OCH ₃	3.81 (3H, s)	56.5 (q)	3.82 (3H, s)	56.5 (q)
3'-OCH ₃	3.68 (3H, s)	52.3 (q)	(- , -,	(1)
OCH ₂ CH ₂ CH ₂ CH ₃	(- , -,	(1)	4.08 (2H, t, 6.5)	65.3 (t)
OCH ₂ CH ₂ CH ₂ CH ₃			1.58 (2H, m)	30.4 (t)
OCH ₂ CH ₂ CH ₂ CH ₃			1.33 (2H, m)	19.0 (t)
OCH ₂ CH ₂ CH ₂ CH ₃			0.90 (3H, t, 7.2)	13.6 (q)

Table 1. ¹H and ¹³C NMR spectral data of **1** and **2** in CDCl₃ (δ in ppm, J in Hz).

3.3.3 5-Hydroxyhexyl 2-hydroxypropanoate (3)

Colorless oil; $[\alpha]_D^{20} - 20.1$ (c = 0.49, acetone). IR (KBr) ν_{max} : 3389 (OH), 1736 (C=O), 1459, 1214, 1134 cm⁻¹; ¹H NMR (500 MHz, CDCl₃, 20°C, TMS): δ 4.26 (1H, m, H-2), 4.17 (2H, m, H-1'), 3.79 (1H, m, H-5'), 1.68 (2H, m, H-2'), 1.47 (2H, m, H-4'), 1.45 (2H, m, H-3'), 1.40 (3H, d, J = 6.2 Hz, H-3), 1.18 (3H, d, J = 6.2 Hz, H-6'). ¹³C NMR (100 MHz, CDCl₃, 20°C, TMS): δ 20.4 (q, C-3), 22.0 (t, C-3'), 23.6 (q, C-6'), 28.5 (t, C-2'), 38.6 (t, C-4'), 65.5 (t, C-1'), 66.7 (d, C-2), 67.8 (d, C-5'), 175.8 (s, C-1). HR MS ((-)ESI): m/z 189.1127 [M-H]⁻ (calcd for C₉H₁₇O₄, 189.1126).

Acknowledgements

This work was financially supported by the National Basic Research Program of China (2009CB522300), and MOST (2009ZX09501-029; 2009ZX09501-013).

References

[1] W. McClatchey, *Integr Cancer Ther.* 1, 110 (2002).

- [2] Y. Deng, Y.W. Chin, H. Chai, W.J. Keller, and A.D. Kinghorn, J. Nat. Prod. 70, 2049 (2007).
- [3] S.M. Sang, K. He, G.M. Liu, N.Q. Zhu, M.F. Wang, J.W. Jhoo, Q.Y. Zheng, Z.G. Dong, G. Ghai, R.T. Rosen, and C.T. Ho, *Tetrahedron Lett.* 42, 1823 (2001).
- [4] S.M. Sang, K. He, G.M. Liu, N.Q. Zhu, X.F. Cheng, M.F. Wang, Q.Y. Zheng, Z.G. Dong, G. Ghai, R.T. Rosen, and C.T. Ho, Org. Lett. 3, 1307 (2001).
- [5] T. Kanchanapoom, R. Kasai, and K. Yamasaki, *Phytochemistry* 59, 551 (2002).
- [6] K. Kamiya, Y. Tanaka, H. Endang, M. Umar, and T. Satake, *Chem. Pharm. Bull.* 53, 1597 (2005).
- [7] C.A. Hornick, A. Myers, H. Sadowska-Krowicka, C.T. Anthony, and E.A. Woltering, *Angiogenesis* 6, 143 (2003).
- [8] R.W. Li, S.P. Myers, D.N. Leach, G.D. Lin, and G. Leach, J. Ethnopharmacol. 85, 25 (2003).
- [9] S.X. Deng, A.K. Palu, B.J. West, C.X. Su, B.N. Zhou, and J.C. Jensen, *J. Nat. Prod.* 70, 859 (2007).
- [10] T. Akihisa, K. Matsumoto, H. Tokuda, K. Yasukawa, K. Seino, K. Nakamoto, H. Kuninaga, T. Suzuki, and Y. Kimura, J. Nat. Prod. 70, 754 (2007).
- [11] T. Hiwasa, Y. Arase, Z. Chen, K. Kita, K. Umezawa, H. Ito, and N. Suzuki, *FEBS Lett.* 444, 1730 (1999).